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# PRELIMINARY PHYTOCHEMICAL SCREENING OF *Pimenta racemosa var. racemosa* (MYRTACEAE) FROM TÁCHIRA – VENEZUELA

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#### Abstract

*Pimenta racemosa* var. *racemosa* (Mill.) J. W. Moore, commonly known as Bay-rum, Malagueta, Pepita species and pepper, is a Caribbean tree, belonging to Myrtaceae family, with a special interest in their leaves for the production of essential oil well known as "Bay-Rum". According to literature consulted there are no reports either for phytochemical or pharmacological activities with respect to extracts of this species. The aim of this study was to evaluate extracts from aerial parts (leaves, fruits, stems and bark) and roots of *P. racemosa* var. *racemosa* collected in April 2012, in order to perform a preliminary phytochemical screening. All extracts were obtained by maceration, and phytochemical screening was carried out by simple colorimetric and TLC techniques. Results indicated mainly the presence of terpenoids, steroids, saponins, phenolic nuclei, glycosides, tannins, and quinines. This information no only supports the use of this plant in traditional medicine but might be useful for further chemical and biological investigations.

Keywords: phytochemical screening, Pimenta racemosa var. racemosa, terpenoids, steroids, phenolic nuclei.

### Introduction

Genus Pimenta belongs to family Myrtaceae, comprises 21 species including several varieties, is typical from tropical America (1). In Venezuela is only represented by P. racemosa (Mill.) JW Moore (P. acris Kostel), and is distributed in Federal District, Falcón, Lara, Merida, Nueva Esparta, Sucre, Tachira and Zulia states (2), commonly known as: Bay-rum, Malagueta, Pepita and pepper species, is cultivated as an ornamental (3), and used in folk medicine for the treatment of various diseases. such as fever, rheumatism, toothache, abdominal pain, pneumonia, colds (4-5), pectoral angina, diarrhea, incontinence, stroke, anti-inflammatory and analgesic properties. Furthermore, its dry wood is used in fence posts because of their resistance against termites (5).

Among pharmacological effects reported for different *Pimenta* species include: insect repellent (6), antidermatophytic (7), antinociceptive (4, 8-9), anti-inflammatory (4, 10), antipyretic (9), antimicrobial (11-19), antioxidant (20-26), anti-cancer (23), antimutagenic (22), anti-hemorrhagic bleeding and cobra venom (27), central nervous system depressant (8, 28-29), hypoglycemic (30), hypotensive (8, 28-29), inhibitor of histone acetyl transferase enzyme (31), and inhibitor of enzyme histidine decarboxylase (32).

*Pimenta* genus chemistry has also been explored for leaves and fruits and phytochemical studies have primarily led to the isolation of various tannins, phenolic compounds, flavonoids, one triterpene saponin and a structural variety of volatile components such as monoterpenes, sesquiterpenes and phenylpropenes (33).

P. racemosa var. racemosa, is an aromatic tree of tropical origin. To date, studies of this species have been focused on the leaves due to the content of volatile essences, once distilled, are used in the manufacture of cosmetics, especially in formulations such as after shave lotions, soaps, colonies and hair treatments (34-35), however to date, literature has not reported studies on phytochemicals or pharmacological activities with respect to extracts of this species, thus, it is expected that extracts might be a source of similar secondary metabolites reported for gender. The aim of this study is to confirm the presence or absence of these metabolites by performing a detailed phytochemical screening that might be useful in further phytochemical investigations.

### **Materials and methods**

### **Plant material**

Leaves, fruits, stems (fine and coarse), bark and roots of P. racemosa var. racemosa were collected in April 2012, near Sector "Los Corredores de la Palmita" Junín Municipality, Rubio town, located at southwestern Táchira state, Venezuela, altitude 859 m.s.n.m. Botanical identification was carried out by Dr. Leslie R. Landrum, Herbarium Curator, School of Life Sciences, Arizona State University, USA. Specimens collected in field are sheltered in MERF Herbarium of the Faculty of Pharmacy and Biomedical Sciences, University of Los Andes (BC-01 code), Venezuela and in ASU Arizona Herbarium of Arizona State University (ASU0075448 code), USA.

### Extraction

Plant material (leaves, fruit, thin stems, thick stems, bark and roots) was dried in an oven with air circulation at 40 °C for 5 days , then was pulverized to yield 4615g leaves (L), 566g fruits (F), 8160g thin stems (TS), 16105g thick stems (TSt), 5572g cortex (C) and 3905g roots (R). All extracts were extracted by soaking with isopropyl alcohol (1:4 m/v) in 4 periods of 21 days each, thereby achieving an exhaustive extraction. Extracts obtained were filtered using filter paper (Double Rings No. 203) and concentrated under reduced pressure at 30°C to remove the remaining isopropyl alcohol. After concentration every extract was weighted yielding 604.11 g (L), 58.17 g (F), 434,82 g (TS), 508,15 g (TSt), 265,44 g (C) y 109,52 g (R).

# **Phytochemical screening**

Crude extracts of L, F, TS, TSt, C and R of *P. racemosa* var. *racemosa* were phytochemically evaluated to determine the presence of chemical constituents using standard procedures, which are described below:

<u>Testing for Alkaloids:</u> each extract (10 mg) was dissolved in 2 mL of hydrochloric acid 5%, after mixing and filtered, three aliquots were taken. Drops of Wagner, Mayer and Dragendorff reagents were added to each. A red-brown precipitate (Wagner), yellowish white precipitate (Mayer) and red-orange precipitate (Dragendorff) indicated the presence of such metabolites (36).

Testing for Coumarins: 10 mg of each sample were

added to 0.5 mL of ethanol along with 2 drops of concentrated ammonium hydroxide. If examination under UV light at a wavelength of 365nm shows the presence of blue or green fluorescence might be indicative of a positive result (36).

<u>Testing for Glycosides:</u> 10 mg of each extract were dissolved in 1 mL of distilled water followed by 5 drops of aqueous sodium hydroxide. A yellow colour indicated the presence of glycosides in the extract (41).

# Testing for Cardiotonic glycosides:

- Keller-Killiani reaction: 10 mg of each extract were dissolved in Keller's reagent and then 5 drops of concentrated sulfuric acid were added. The occurrence of brown ring between the two phases formed is indicative of deoxy glycosides nuclei presence. (36)

- Legal Reaction: 3 drops of pyridine, 1 drop of sodium nitroprusside solution (aqueous) 5% and 3 drops of 2N sodium hydroxide were added to 10 mg of each extract. Intense red coloration indicated the presence of cardenolides, or  $\alpha$ , $\beta$ -unsaturated lactones nuclei (36).

<u>Testing for Flavonoids:</u> Shinoda, Pew's, 10 % sodium hydroxide and 2-aminoethyl diphenyl borate assays were performed in order to determine the presence of these metabolites.

- Shinoda test: 1 mL of absolute ethanol and 3 drops of concentrated hydrochloric acid were added to 10 drops of diluted extract in isopropyl alcohol. Formation of red color indicated the presence of aurones and chalconnes. In cases where no colour change was observed, pieces of metallic magnesium were placed. The formation of orange, red or magenta coloration indicated the presence of flavones, flavonols and flavones, respectively (36).

- Pew's test: 2 mg of zinc powder and 5 drops of 5N hydrochloric acid were added to 1 mL of each diluted extract in isopropyl alcohol. The presence of red, pink or coffee color indicated the existence of dihyflavones, flavonones, and dihydrochalcones, respectively (36).

- Test with 10% sodium hydroxide: 3 drops of sodium hydroxide 10% were added to 1 mL of diluted extract in isopropyl alcohol. Formation of yellow-red, coffee-orange, purple-red or blue coloration indicated the presence of xanthones and/or flavones, flavonols, chalcones and anthocyanins, respectively (36). - Test with 2-aminoethyl diphenylborate: Methanolic solutions of each extract were analyzed by thin layer chromatography (TLC) on silica gel 60 F254 plates, at room temperature. Solvent system was chloroformmethanol formic acid-water (3:6:1:1). Visualization was performed under UV-Vis light (350nm) before and after staining with 1% methanolic 2-aminoethyl diphenylborate reagent. Appearance of yellow fluorescence, after spraying with reagent was indicative of flavonoids nuclei (38-39).

<u>Testing for steroids and/or triterpenoids:</u> Salkowski, Rosenthaler and Lieberman-Bouchard assays were carried out (36).

- Salkowski test: 2 mL of chloroform and 1 mL of concentrated sulfuric acid were added to 10 drops of the extract dissolved in isopropyl alcohol, slowly until double phase formation. The presence of a reddishbrown color in the middle layer was indicative of steroidal ring.

- Rosenthaler test: 3 drops of Rosenthaler reagent and 2 drops of concentrated sulfuric acid were added to 2 mL of the extract dissolved in isopropyl alcohol. Formation of violet color in the middle layer was indicative of terpenoids.

- Lieberman-Bouchard test: 1 mL of anhydrous acetic acid and 3 drops of concentrated sulfuric acid were added to 2 mL of the extract dissolved in isopropyl alcohol. After 5 minutes a blue-green color middle layer was indicative of sterols, but a pink, red, magenta or violet color revealed the presence of terpenoids.

<u>Testing for Quinones and Anthraquinones</u>: Presence of such metabolites was determined by Borntraguer, ammonium hydroxide, concentrated sulfuric acid (36) and benzene assays (40).

-Borntraguer test: 10 mg of each extract were dissolved in 3 mL of distilled water and filtrated. After filtration 3 mL of 5% potassium hydroxide solution were added to each. The mixture was heated to boiling for 3 minutes. Alkaline solution was allowed to cool down and then extracted with 3 mL of chloroform. Organic phase was separated and shaken with 2 mL of 5% potassium hydroxide solution. Occurrence of red color in alkaline phase indicated the presence of quinones. Those samples showing yellow colour with green fluorescence where treated with one drop of 6% hydrogen peroxide, formation of red colour was considered positive for anthrones derivatives. - Test with ammonium hydroxide: One drop of concentrated ammonium hydroxide was added to 10 mg of each extract, previously dissolved in isopropyl alcohol. After two minutes, formation of red colour indicated the presence of anthraquinone.

- Test with sulfuric acid: One drop of concentrated sulfuric acid was added to 10 mg of each extract dissolved in isopropyl alcohol. Formation of red color indicated the presence of quinones.

- Test with benzene: 1 mL of benzene was added to 10 mg of each extract dissolved in isopropyl alcohol, followed by stirring and filtration. 0.5 mL of 10% ammonia solution was added to the filtrate. This mixture was shaken; formation of pink, red or violet colour in the ammoniacal (lower) phase indicated the presence of anthraquinones.

Testing for Tannins: 100 mg of each extract were dissolved in 10 mL of ethanol, and extracted with 25 mL of distilled water in boiling during 15 minutes. Once allowed to fresh at room temperature, 0.2 mL of 10% sodium chloride solution were added to mixture and filtered. The filtrate was divided into four equal portions in test tubes. 5 drops of 1% gelatin solution were added to first portion, 5 drops of gelatin - salt solution (1% gelatin + 10% salt) were added to second portion, 4 drops of 10% ferric chloride solution were added to third portion, and 3 drops of 1% potassium ferricyanide solution were added to fourth portion. Precipitation observed after addition of either second or third reagent was indicative of the presence of tannins. Those samples showing gravish-black or black-blue colour after addition of third reagent indicated the presence of tannins with catechol or pyrogallol nuclei, respectively. Samples turning to blue colour after fourth reagent revealed the presence of phenolic compounds (41).

<u>Testing for Phenols:</u> 10 mg of each extract were dissolved in 1 mL of ethanol, then 2 mL of distilled water was added followed by 4 drops of ferric chloride aqueous solution 10% w/v. Formation of a blue or green color indicated the presence of phenols (42).

# Testing for Saponins:

- Test Foam height (without sodium bicarbonate): 1 mL of distilled water were added to 10 drops of the extract dissolved in isopropyl alcohol (20 mg/mL) in a test-tube, shaken vigorously to froth, then

allowed to stand for 30 minutes. Saponin content was measured as follows: no froth (absence); froth less than 3 mm high (poor); froth 6 mm high (moderate) and froth greater than 8 mm high (abundant) (36).

- Foam Test (with sodium bicarbonate): 1 mL of distilled water and 1 drop of sodium bicarbonate saturated solution were added to 5 drops of the extract dissolved in isopropyl alcohol (20 mg/mL) in a test-tube and shaken vigorously during 3 minutes. Formation of honeycomb shaped foam indicated the presence of saponins (42).

# **Results and Discussion**

Preliminary phytochemical screening from organic extracts (L, F, TS, TSt, C and R) of *P. racemosa* var. *racemosa* was carried out using various chemical assays in order to identify either the presence or absence of secondary metabolites such as alkaloids, coumarins, phenolic compounds, flavonoids, glycosides, quinones, saponins, tannins, steroids and triterpenoids

The results (Table 1) revealed abundant presence of phenolic nuclei and tannins in all extracts assaved while no alkaloid incidence was observed. In cardiotonic. reference to glycosides steroid. and terpenoids, quinones, flavonoids, these components are present in all extracts evaluated; the only difference might possible be the amounts between these. For instance, deoxy glycosides nuclei were abundant in L extract, moderate in F and C extracts while R extract showed low content. Cardenolides or  $\alpha,\beta$ -unsaturated type lactones were in high concentration in TS, TSt and R extracts, moderate incidence in L and C extracts but minor presence in F extract. Steroids were well evident in L, TS, TSt, C and R extracts while terpenoids were abundant in L and TS respectively. Regarding flavonoids, the presence of flavones or xanthones type components were observed moderately in F extract; dihydrochalcones and/or flavonols in R, L, TSt and C. Quinones showed to be moderated in F, TS, TSt, C and R extracts. Moreover, saponins were present in five of the extracts assayed (L, TS, TSt, C, R), being R extract the most abundant between these followed by TSt.

Anthraquinone nuclei was observed abundant in TS, TSt extracts but moderated in R and C, respectively

According to previous investigations *P. dioica* and *P. pseudocaryophyllus* are composed mainly by flavonoids (27, 32, 44), tannins (23, 27, 44) and terpenes present as essential oils (7, 16,17, 43). Moreover, their fruits revealed the presence of phenylpropanoids (20), tannins (21) and phenolic compounds (51). *P. racemosa* var. *ozua* leaves revealed mainly the presence of lupeol (10) but leaves of *P. adenoclada* (46), *P. haitiensis* (47), *P. jamaicensis* (48), *P. obscura* (49), *P. racemosa* var. *grisea* (14, 45), *P. racemosa* var. *hispaniolensis* (45), *P. racemosa* var. *terebinthina* (14) showed only terpene type components.

Comparing results from present investigation to previous reported for *P. dioica* and *P. pseudocaryophyllus*, occurrence of flavonoids, phenolic compounds, tannins and saponins were corroborated, however, other secondary metabolites such as glycosides, quinones, steroids and terpenoids were also observed in all extracts assayed.

Ever since no published works have been found regarding phytochemical studies on different parts of *Pimenta racemosa* var. *racemosa*, present investigation is consider as a qualitative type examination that reveals information about several possible nucleus present on this species that might be useful as reference for further investigations.

On the other hand, secondary metabolites observed in all extracts assayed in present investigation might be associated to possible biological activities. According to references consulted flavonoids have been associated to antimicrobial, anticancer, antiinflammatory and antioxidant activities (42). Tannins have revealed antiviral, anti-inflammatory and antitumor properties (42). Saponins are used mainly as mild detergent, however has exhibited antioxidant, antifungal, anticancer among other activities (42). Triterpenes and steroids act as antiseptic, antibacterial and antifungals (52), quinones are used as anthelmintic, have laxative effects and are used in the treatment of fever, gonorrhea, leprosy and menstrual cramps (37). Polyphenols exhibit a wide range of biological activities, including anti-cancer, anti-inflammatory, antihypertensive, estrogenic, antioxidant and to prevent cardiovascular diseases (53).

### Conclusions

Concerning *Pimenta* genus, flavonoids, tannins, phenolic compounds, saponins, and glycosides are

the most common metabolites reported.

Specie assayed in present investigation showed to be consistent to previous reports. However, coumarins, glycosides and  $\alpha$ , $\beta$ -unsaturated ketones were also observed in several parts of the plant. According to references consulted for *P. racemosa* var. *racemosa* only essential oil composition has been published to date.

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Secondary metabolite	Testing	L	F	TS	TSt	С	R
Alkaloids	Wagner, Mayer, Dragendorff	-	-	-	-	-	-
Coumarins	Fluorescence at 365nm	-	-	-	-	-	+
Glycosides	Reaction with NaOH (ac)	-	+	-	-	-	-
Glycosides cardiotonic	Keller – Killiani (deoxy glycosides nuclei)	+++	++	+	+	++	+
	Reaction of Legal		+	+++	+++	++	+++
	(cardenolides, or $\alpha$ , $\beta$ –unsaturated lactones nuclei)	++					
Flavonoids	Shinoda (aurones, chalconnes, flavones, flavonols and						
	flavones)	-	-	-	-	-	-
	Pew's (dihyflavones, flavonones, and dihydrochalcones)	-	-	+	-	++	+
	NaOH 10% (xanthones and/or flavones, flavonols,	++	++	+	++	++	++
	chalcones and anthocyanins)						
	TLC: 2-aminoetil difenil Borato	++	+++	-	-	-	-
Steroids y/o Terpenoids	Lieberman Bouchard	+++	+	+++	++	++	+++
	Rosenthaler (Terpenoids)	+++	+	+++	+++	+	++
	Salkowskí (Steroids)	+++	+	+++	+++	+++	+++
	TLC: Komarowsky	+++	+	++	++	+++	++
Phenols	FeCl <sub>3</sub> 10%	+++	+++	+++	+++	+++	+++
Quinones	C <sub>6</sub> H <sub>6 (conc.)</sub> (anthraquinone)	-	-	-	-	-	-
	NH <sub>4</sub> OH <sub>(conc.)</sub> (anthraquinone)	-	-	++	++	++	++
	Borntraguer (quinone or anthrones derivatives)	-	-	-	-	-	-
	H <sub>2</sub> SO <sub>4 (conc.)</sub> (quinone)	+	++	++	+++	++	++
Saponins	Foam height	+	-	+	++	+	+++
	Foam (with sodium bicarbonate)	+	-	++	+	+++	+++
Tannins	1% Gelatin, Gelatin-Salt solution (1% gelatin + 10% salt),						
	Potassium ferricyanide 1%	-	-	-	-	-	-
	FeCl <sub>3</sub> 10% (catechol or pyrogallol nuclei)	+++	+++	+++	+++	+++	+++

**Table 1.** Phytochemical screening of alcoholic extracts achieved from different parts of *P. racemosa* var. racemosa.

**Key:** - Absence, + Poor, ++ Moderate, +++ Abundant, L: alcoholic extract of leaves, F: alcoholic extract of fruits, TS: alcoholic extract of fine branches, TSt: alcoholic extract of large branches, C: alcoholic extract of cortex and R: alcoholic extract of roots.