

SYNTHESIS AND ANTI DIABETIC ACTIVITY OF SOME NEW 2-(SUBSTITUTED PHENYL)-4H-CHROMEN-4-ONE DERIVATIVES

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Summary

A new series of 2-(substituted phenyl) chromen-4-one derivatives were synthesized. All the synthesized compounds were purified and R_f values were checked. The structures of these compounds were confirmed by IR, NMR, and CHN analysis. These compounds were subjected to anti diabetic activity. All the compounds have shown promising anti diabetic activity.

Key words: Chromone derivatives, Anti diabetic activity, CHN analysis

Introduction

A number of natural and synthetic Benzopyrone derivatives have been reported to exert notably antimicrobial, antitubercular and antifungal activity. ¹⁻³ Benzopyrone having Chromone (γ -Benzopyrone) moiety are associated with interesting physiological activities such as anti microbial anti tubercular anti inflammatory anti diabetic antiviral anticancer. ⁴⁻¹²

In view of these observations and our interest in the synthesis of biologically active bi heterocycle possessing Chromone nucleus, we have modified 5-amino-6-hydroxy-2-phenyl-4H-Chromen-4-one V by reacting it with thiourea to yield 2-amino-7-phenyl chromen (5,6-d) imidazol-9 (3H)-one A₁ for its versatile biological activity. Compound V was treated with urea to yield 7-phenyl-1H-Chromeno (5,6-d) oxazole-2, 9-dione A₂. Compound v was treated with chloroacetyl chloride to get 8-phenyl chromeno (6,5-b) (1,4) oxazine-2, 10 (1H, 3H)-dione A₃. Compound v was treated with substituted aldehydes to yield 2-(4-substituted phenyl)-7-phenyl-9H-chromeno (5,6) oxazol-9-one A₄ and A₅ to explore activities associated with this nucleus and screened them for anti-inflammatory activity. The unique structure of compound V has facilitated for getting compound A₁ to A₅, which would possess the promising anti-inflammatory activity.

Material and Methods

Antidiabetic Activity¹³

The compounds synthesized during the present work were subjected to anti diabetic activity. Using wistar albino rats.

Method: Alloxan induced tail-tipping method. The animals were selected of either sex weighing between 150-200 gms of body weight. The experimental animals kept on fasting for 24 hrs, and then known quantity of Alloxan was induced to increase the body glucose level. The experimental compounds were thus induced and the glucose level was measured by tail tipping method-using glucometer. The results were recorded and calibrated

Experimental

Melting points were determined in open capillary method and are uncorrected. Purity of the compound was checked on Silica gel TLC plates. IR spectra were recorded on Thermo Nicolet IR 200 spectrophotometer using KBr disc method. ¹HNMR spectra were recorded on Bruker AMX-400, DMSO d₆ as internal standard. Combustion analyses were found to be within the limits of permissible errors.

Synthesis of 1-(2,5dihydroxy phenyl)-3-phenyl-1, 3-propanedione II¹⁴

54 gm (0.15 mole) of 2,5-dibenzoyloxy acetophenone I was dissolved in 325 ml of dry pyridine and was heated to 50 °C in a flask. Further 7.3 g (0.13 moles) of finely powdered potassium hydroxide were added with stirring. Stirring was continued till the solid separated out. The reaction mixture cooled to room temp. Acidified with 187 ml of 10% acetic acid solution and the solid separated out was filtered, dried and recrystallized from rectified spirit to get the product (24.4 g), yield 64%.

Synthesis of 6-hydroxy-2-phenyl-4H-chromen-4-one III.¹⁴

25.6 g (0.1 mole) of 1-(2,5-dihydroxyphenyl)-3-phenyl-1, 3-propanedione II was dissolved in 75 ml of glacial acetic acid in a 500 ml of round bottom flask and 4 ml of conc. Sulphuric acid was added with constant stirring. The reaction mixture was refluxed for 2h and cooled. The contents of the flask were poured in to beaker containing crushed ice. The solid separated out was filtered and it was recrystallized from n-hexane to get the (17.2 g), yield 68%, m.p 240-42⁰.

Synthesis of 6-hydroxy-5-nitro-2-phenyl-4h-chromen-4-one IV.¹⁵

A mixture of 6-hydroxy-2-phenyl-4h-chromen-4-one III (5.1 g, 0.02 moles) and concentrated sulphuric acid (30 ml) was stirred at 0⁰ for 30 min. Then a mixture of conc. Nitric acid (1.5 ml) and sulphuric acid (5 ml 98%) were added. The temperature was kept at 0-5⁰ during the period of addition, and the mixture was then continuously stirred for 2 h at 5⁰. The reaction mixture was poured into ice cold water the precipitate formed was filtered dried and recrystallized from acetone to get the product (4.02 g), yield 78%, m.p. 210-12⁰.

Synthesis of 5-amino-6-hydroxy-2-phenyl-4H chromen 4- one¹⁴

Iron powder (8g) was added portion wise to a hot mixture of 6-hydroxy-5-nitro-2-phenyl-4h-chromen-4-one IV (5.66 g, 0.02 moles) in ethyl alcohol (20 ml) and conc. Hydrochloric acid (30 ml) at reflux temperature. After completion of the addition, the refluxing was continued for 6h. Upon cooling a white precipitate formed was filtered off washed with water, dried and recrystallized from methanol to get product (3.1g) yield 56%, m.p. 180-82⁰.

Synthesis of 2-amino-7-phenyl chromen (5, 6-d) imidazol-9- (3H)-one A₁.

A mixture of V (0.5 g 0.002 mole) and thiourea (2.28 g, 0.03 mole) was heated at 130-140⁰ for 15 min the reaction mixture melted and re-solidified, treated with hot water, filtered off and recrystallized from ethanol to get the product (0.3 g), yield 62%, m.p. 194-95⁰.

Synthesis of 7-phenyl-1H-chromeno (5, 6-d) oxazole-2, 9-dione A₂

A mixture of V (0.5 g, 0.002 mole) and urea (1.8 g, 0.03 mole) was heated at 100⁰ for 15 min, the reaction mixture melted and re-solidified, treated with hot water, filtered off and recrystallized from ethanol to get the product (0.24 g), yield 48%, m.p. 208-09⁰.

Synthesis of 8-phenyl chromeno (6, 5-b) (1, 4) oxazine-2, 10 (1H, 3H)-Dione A₃

A mixture of V (0.5 g, 0.002 moles), chloroacetyl chloride (0.17 ml, 0.002 moles) and anhydrous potassium carbonate (0.5g) in dry acetone (20 ml) was refluxed for 3h, cooled then poured into ice-cold water. The precipitate formed was filtered off and recrystallized from ethanol, yield 57% (0.28 g), m.p. 231-32⁰.

Synthesis of 2-(4-substituted phenyl)-7-phenyl-9H-chromeno (5, 6) oxazol-9-one A₄, A₅

To a solution of V (0.5g, 0.002 mole) in glacial acetic acid and the appropriate aldehydes namely nitro benzaldehyde and p-chloro benzaldehyde (0.002 mole) was refluxed for 15 hrs, cooled and poured into ice-cold water. The precipitate formed was filtered off and recrystallized from pet. Ether, A₄: yield 39% (0.19 g), m.p. 178-79⁰, A₅: yield 41% (0.21g), m.p. 136-37⁰.

Results and Conclusions

A new series of 2-(substituted phenyl) chromen-4-one derivatives were synthesized and the structures of these compounds were confirmed by IR, NMR, and CHN analysis. These compounds were subjected to anti diabetic activity.

Compound A₃, A₄, A₅ have given excellent anti diabetic against Glibenclamide as a standard drugs. With the suitable molecular modification of these synthesized compounds, they may prove as a potent anti diabetic agent.

SPECTRAL DATA

A₁ : IR (KBr) cm^{-1} : 3232(-NH str.), 3055 (Ar-CH str.), 1652 (C=O pyrone), 1500 (Ar-C-C str.), 1331 (C-O str.). ¹H NMR (d ppm): 11.89 (s, 1H, NH), 7.0-7.63 (m, 8H, Ar-H), 6.56 (d, 2H, NH₂)

A₂ : IR (KBr) cm^{-1} : 3238(-NH str.), 3046 (Ar-CH str.), 1689 (C=O pyrone), 1501 (Ar-C-C str.), 1331 (C-O str.). ¹H NMR (d ppm): 11.59 (s, 1H, NH), 7.52-7.62 (m, 8H, Ar-H).

A₃: IR (KBr) cm^{-1} : 3238(-NH str.), 3085(Ar-CH str.), 1651 (C=O pyrone), 1501 (Ar-C-C str.), 1301 (C-O str.). ¹H NMR (d ppm): 10.73 (s, 1H, NH), 7.0 -7.63 (m, 8H, Ar-H), 6.96 (d, 2H, CH₂).

A₄: IR (KBr) cm^{-1} :3073 (Ar-CH str.), 1661 (C=O pyrone), 1516 (Ar-C-C str.), 1280 (NO₂), 679 (-Cl str.). ¹H NMR (d ppm): 6.94-7.64 (m, 8H, Ar-H).

A₅: IR (KBr) cm^{-1} :3062 (Ar-CH str.), 1682 (C=O pyrone), 1588(Ar-C-C str.), 1452 (C-N str.), 1360 (C-O str.), 1263 (-CN str.). ¹H NMR (d ppm): 6.94-7.64 (m, 8H, Ar-H).

SCHEME

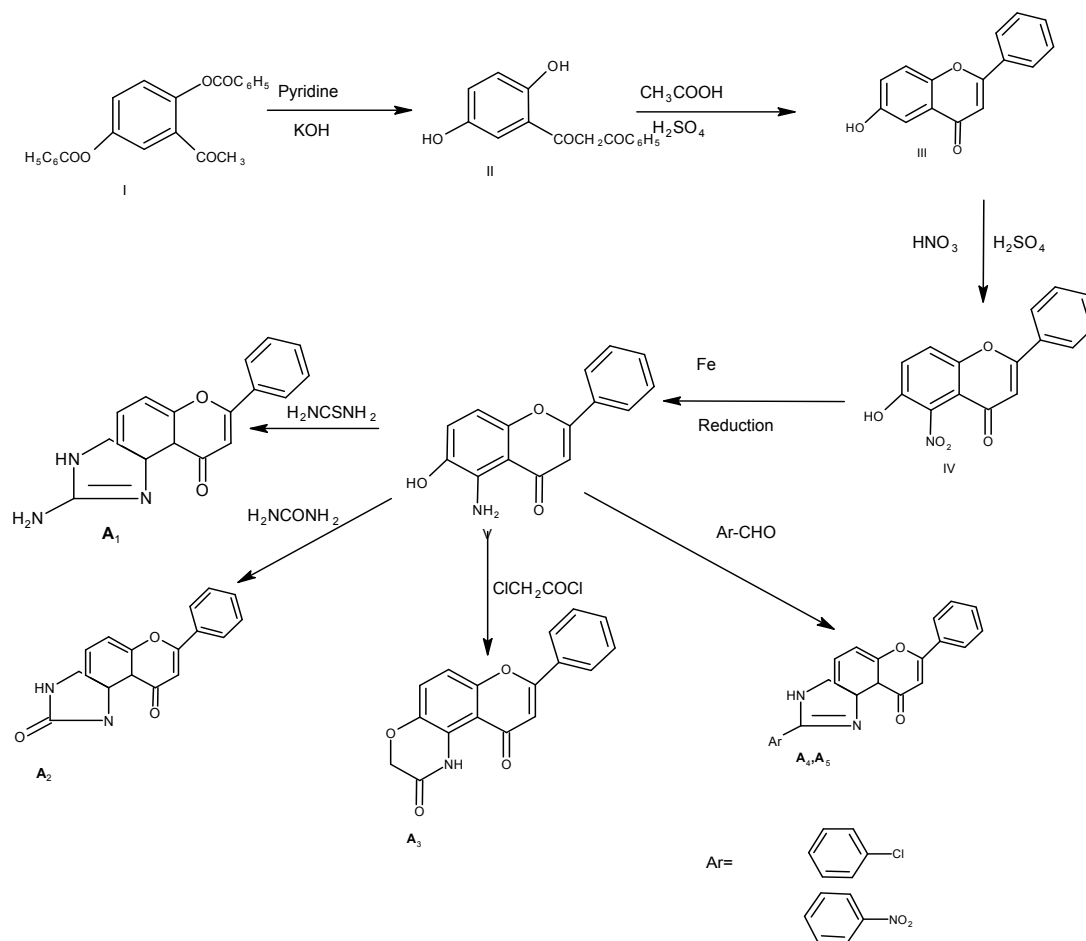


Table no 1: Analytical data of synthesized compounds (A₁-A₅)

Compd.	Mol. Formula	Mol. Wt.	Yield %	m.p. ° C	Elemental analyses Calcd.(Found)		
					C	H	N
A ₁	C ₁₆ H ₁₁ N ₃ O ₂	277	62	194-195	69.3 (69.52)	4.0 (3.92)	15.15 (15.21)
A ₂	C ₁₆ H ₉ N ₃ O ₄	279	48	208-209	68.82	3.25	5.02
A ₃	C ₁₇ H ₁₁ NO ₄	293	57	231-232	69.62	3.48	4.78
A ₄	C ₂₂ H ₁₂ NO ₃ Cl	373	39	178-179	70.69 (70.42)	3.24 (3.58)	3.75 (4.00)
A ₅	C ₁₆ H ₁₂ N ₂ O ₅	384	41	136-137	68.75	3.15(3.50)	7.29

The combustion analysis of compounds synthesized is within the limits of permissible errors.

Table no.2: Anti diabetic activity of synthesized compound

Drug	Blood glucose level mg/dl (Mean ± SE)			
	0 h	1 h	3 h	6 h
Control	123.3± 6.00	120.7± 5.54	122.3± 5.81	123.0± 6.4
Glibenclamide	385.8±21.37	230.8±12.35**	156.8±10.87**	93.4±4.9**
A ₁	387.5±19.27	250.7±14.19	200.1±13.09	125.5±7.1
A ₂	368.3±17.57	243.9±13.19	193.7±12.17	135.6±7.2
A ₃	379.2±20.29	233.3±12.57**	154.9±10.25**	95.3±4.2**
A ₄	383.1±19.47	220.8±11.47**	149.7±9.57**	93.5±4.5**
A ₅	385.7±23.29	219.7±11.09**	145.3±9.01**	91.3±3.9**

p<0.05 *, p<0.01**, Statistical ANOVA followed by Dunnet 't' test when compared with 0 hour reading. The drugs have shown significant anti diabetic activity. The drugs have shown moderate anti diabetic activity on oral administration.

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