Hepatoprotective and Antioxidant Activities of *Momordica Cymbalaria* Hook. F.

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**Summary**

Methanolic extract of *Momordica cymbalaria* Hook.F. (MEMC) was studied against carbon tetrachloride (CCl₄) induced hepatic injury in albino rats and mice. In addition, antioxidant activity was studied by in vitro models. Pre-treatment with MEMC reduced biochemical markers of hepatic injury like serum glutamate pyruvate transaminase (SGPT), serum glutamate oxaloacetate transaminase (SGOT), alkaline phosphatase (ALP), bilirubin, cholesterol, high density lipoprotein (HDL), and tissue glutathione (GSH) levels. Similarly pretreatment with MEMC reduced the CCl₄ induced elevation in the pentobarbitone sleeping time. Histopathological observations also revealed that pretreatment with MEMC protected the animals from CCl₄ induced liver damage. MEMC demonstrated dose dependant reduction in the in vitro and in vivo peroxidation induced by CCl₄. In addition it showed dose dependant free radical scavenging activity. The amounts of total phenolic compounds were also determined in this study. The results obtained in the present study indicate that the MEMC can be a potential source of natural antioxidant and hepatoprotective activity.

**Key words:** MEMC, *Momordica cymbalaria*, Antioxidant activity, hepatoprotective, DPPH assay, CCl₄

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Introduction

Liver is the most important organ, which plays a pivotal role in regulating various physiological processes in the body. It is involved in several vital functions, such as metabolism, secretion and storage. It has great capacity to detoxicate toxic substances and synthesize useful principles. Therefore, damage to the liver inflicted by hepatotoxic agents is of grave consequences (1, 2). Liver diseases are mainly caused by toxic chemicals, excess consumption of alcohol, infections and autoimmune disorders. Most of the hepatotoxic chemicals damage liver cells mainly by inducing lipid peroxidation and other oxidative damages (3-5). In spite of tremendous advances in modern medicine, there are no effective drugs available that stimulate liver function, offer protection to the liver from damage or help to regenerate hepatic cells (6). In absence of reliable liver-protective drugs in modern medicine, there are a number of medicinal preparations in Ayurveda recommended for the treatment of liver disorders and their usage is in vogue since centuries (7).

*Momordica cymbalaria* Hook. F. belongs to the Cucurbitaceae family. The plant is a perennial herbaceous climber either allowed to trail on the ground or to climb on supports with the aid of tendrils. It is found in the south Indian states of Andhra Pradesh, Karnataka, Madhya Pradesh, Maharashtra and Tamil Nadu as a weed. The plant is allowed to grow along bunds (boundary of fields), fences and even in the fields for the sake of fruits. However, no regular cultivation is practiced. The plant has a tuberous root, which helps to maintain perennial habits, pubescent or sub glabrous, i.e., the plants dry and disappear at the end of the season. The tubers remain in the soil and emerge in the next season. The plant has a monocious stem and is very slender. The leaves are oblicular or reiform with a deeply cordate base. Flowers are unisexual. The male flower peduncle is 5–30 mm long, filiform, puberulaus, ebracteate with 2–5 flowers in racemes with a pale yellow corolla and two stamens for each flower. The female flower is solitary on a peduncle of 28 mm length. The fruits are 20–25 mm long, pyriform with 8 sharp ridges, 24×15 mm attenuated at the apex and with the base narrowed into the curved peduncle, which is fleshy, dark green and ribbed. The seeds are 4.6 mm long, ovoid shaped, smooth and shiny. Flowering occurs during October; fruits are harvested from November to January. The yield of each plant is 1.25 to 1.5 kg. The tender fruits closely resemble those of a small variety of bitter gourd Athalakkai is used as a vegetable by the rural people of South Tamil Nadu and North Karnataka, India [8] The phytochemicals reported in this plants are tannins, alkaloids, phenols, proteins, amino acids (9), Vitamin C, carbohydrate and B-Carotene (8). The fruits of this plant reported anti diabetic and antihyperlipidemic activities. The tubers were reported as antiovulatory activity. (9, 10)

Furthermore, literature survey of *M. cymbalaria* revealed that no researcher has yet reported hepatoprotective and antioxidant activities of this plant. Therefore, it is worth conducting an investigation on the hepatoprotective and antioxidant methanolic extract of *M. cymbalaria* fruits (MEMC).
Materials and Methods

Plant material

The fruits of *Momordica cymbalaria* Hook F. was collected in November 2006 from the Bellary, Karnataka, India. The fruit material was taxonomically identified by the Regional Research Institute, Karnataka, India, and the Voucher specimen RRI/BNG/DSRU/F53/2006-07. The fruits were dried under shade with occasional shifting and then powdered with a mechanical grinder and stored in an airtight container.

Preparation of extract

The powder obtained was subjected to successive soxhlet extraction with the solvents with increasing order of polarity i.e. Pet. Ether (60-80°), Chloroform (59.5-61.5°), Methanol (64.5-65.5°) and water. Yield 3.29, 6.19, 11.70, and 15.71% respectively.

Phytochemical screening

A preliminary phytochemical screening of all extracts carried out as described by Khandelwal K.L (11).

Animals

Albino rats (Wistar) weighing 150-200g and albino mice weighing 20-25g of either sex were used in this study. They were procured from Sri Venkateshwara Enterprises, Bangalore. The animals were acclimatized for one week under laboratory conditions. They were housed in polypropylene cages and maintained at 27°C ± 2°C under 12 hrs dark / light cycle. They were fed with standard rat feed (Gold Mohr Lipton India Ltd.) and water *ad libitum* was provided. The litter in the cages was renewed thrice a week to ensure hygiene and maximum comfort for animals. Ethical clearance for handling the animals was obtained from the Institutional Animals Ethical Committee (RCP/IAEC/07/2006) prior to the beginning of the project work. (129/1999/CPCSEA)

ANTIOXIDANT ACTIVITY

Reducing power

The reducing power of MEMC was determined according to the method of Oyaizu [34]. 10 mg of MEEC extract in 1ml of distilled water was mixed with phosphate buffer (2.5 ml, 0.2 M, pH 6.6) and potassium ferricyanide [K3Fe(CN)6] (2.5 ml, 1%). The mixture was incubated at 500C for 20 min. A portion (2.5 ml) of trichloroacetic acid (10%) was added to the mixture, which was then centrifuged at 3000 g for 10 min. The upper layer of the solution (2.5 ml) was mixed with distilled water (2.5 ml) and FeCl3 (0.5 ml, 0.1%) and the absorbance was measured at 700 nm. Sodium metabisulphite was
used as the reference material. All the tests were performed in triplicate and the results averaged. The percentage decrease in absorbance was calculated. (12)

**Inhibition of Hydroxyl radical**

Hydroxyl radical scavenging activity was measured by studying the competition between deoxyribose and the test compounds (MEMC) for hydroxyl radical generated by Fe3+-Ascorbate-EDTA-H2O2 system (Fenton reaction) according to the method of Kunchandy and Rao (16). The reaction mixture contained in a final volume of 1.0 ml, 100 µl of 2-deoxy-2-ribose (28 mM in KH2PO4-KOH buffer, 20 mM, pH 7.4), 500µmol of the various concentrations of MEMC and the reference compound sodium metabisulphate (25 µg) in KH2PO4-KOH buffer (20 mM, pH 7.4), 200 µl of 1.04 mM EDTA and 200 µM FeCl3 (1:1 v/v), 100 µl of 1.0 mM H2O2 and 100 µl of 1.0 mM ascorbic acid was incubated at 37°C for 1h. 1.0 ml of thiobarbituric acid (1%) and 1.0 ml of trichloroacetic acid (2.8%) were added to the test tubes and incubated at 100°C for 20 min. After cooling, absorbance was measured at 532 nm against a control sample containing deoxyribose and buffer. Reactions were carried out in triplicate. Sodium metabisulphite used as the reference compound. All the tests were performed in triplicate and the results averaged. The percentage decrease in absorbance was calculated. (12)

**Inhibition of DPPH radical**

The free radical scavenging activity of MEMC was measured by 1, 1-diphenyl-2-picryl-hydrazil (DPPH) using the method of Blois (35). 0.1 mM solution of DPPH in methanol was prepared and 1ml of this solution was added to 3 ml of various concentrations of MEMC and the reference compound (5, 10, 25, 50 and 100 µg). After 30 min, absorbance was measured at 517 nm. BHT was used as the reference material. All the tests were performed in triplicate and the graph was plotted with the mean values. The percentage of inhibition was calculated by comparing the absorbance values of the control and test samples (12).

**Inhibition of Nitric oxide radical**

Nitric oxide generated from sodium nitroprusside in aqueous solution at physiological pH interacts with oxygen to produce nitrite ions, which were measured by the Griess reaction (13, 14). The reaction mixture (3 ml) containing sodium nitroprusside (10 mM) in phosphate buffered saline (PBS) and MEMC and the reference compound in different concentrations (5, 10, 25, 50 and 100 µg) were incubated at 25°C for 150 min. Each 30 min, 0.5 ml of the incubated sample was removed and 0.5 ml of the Griess reagent (1% sulphanilamide, 0.1% naphthylethylene diamine dihydrochloride in 2% H3PO4) was added. The absorbance of the chromophore formed was measured at 546 nm. All the tests were performed in triplicate and the results averaged. BHT used as the reference compound. All the tests were performed in triplicate and the results averaged. The percentage decrease in absorbance was calculated. (12)
Inhibition of Superoxide anion radical
Measurement of superoxide anion scavenging activity of MEMC was performed based on the method described by Nishimiki (15) and slightly modified. About 1 ml of nitroblue tetrazolium (NBT) solution containing 156 µM NBT which is dissolved in 1.0 ml of phosphate buffer (100 mM, pH 7.4), 1 ml of NADH solution containing 468 µM of NADH which is dissolved in 1 ml of phosphate buffer (100 mM, pH 7.4) and 0.1 ml of various concentration of MEMC and the reference compounds (5, 10, 25, 50 and 100 µg) were mixed and the reaction started by adding 100 µl of phenazine methosulphate (PMS) solution containing 60 µM of PMS 100 µl of phosphate buffer (100 mM, pH 7.4). The reaction mixture was incubated at 25°C for 5 min and the absorbance at 560 nm was measured against the control samples. Vitamin C used as the reference compound. All the tests were performed in triplicate and the results averaged. The percentage decrease in absorbance was calculated. (12)

Amount of total phenolic compounds
Total phenolic content of MEMC was assessed approximately by using the Folin–Ciocalteau phenol reagent method. To 200 ml of the, sample extracts were added 1.0 ml of Folin–Ciocalteau, reagent and 0.8 ml of sodium carbonate (7.5% w/v),and the contents were mixed and allowed to stand for 30 min. Absorption at 765 nm was measured in a, UV–Vis Spectrophotometer. The total phenolic content was expressed as gallic acid equivalents (GAE) in milligrams, per gram of sample, using a standard curve generated with gallic acid (17)

HEPATOPROTECTIVE ACTIVITY

Acute oral toxicity studies
The acute oral toxicity was performed according to OPPTS following up and down procedure. Colony bred female albino rats Wistar strain (150-200gm) were maintained under controlled animal house condition with access to food and water ad libitum. The limit test carried out first at 5000mg/kg. b.w. All animals were observed for toxic symptoms and mortality for 72 h. (18)

CCl4 -induced hepatotoxicity
In this experiment 24 hr fasted animals were randomly divided in to 6 groups 6 animals each. Animals of group I and II received 1 ml of distilled water, whereas animals of group III, IV, V and VI received silymarin (100mg/kg), MEMC (200, 400, and 600mg/kg) respectively for 5 days orally. Animals of group I received 1ml/kg of liquid paraffin on 2nd and 3rd day, where as animals of group II,III,IV,V and VI received 2 ml
CCl₄: paraffin (1:1) simultaneously on 2nd and 3rd day 30 min after vehicle, silymarin or test extract. Animals were sacrificed under mild ether anesthesia. Blood samples and liver tissue were collected. (19)

Blood samples collected from the above mentioned group of biochemical markers of liver damage like serum glutamate pyruvate transaminase (SGPT) (20), serum glutamate oxalaoacetate transaminase (SGOT) (21), alkaline phosphatase (ALP) (22), serum bilirubin (23), cholesterol (24) and HDL (25).

Liver slices were collected from the above groups of animals were subjected to the determination of tissue glutathione (GSH) level and lipid peroxidation. Tissue GSH levels were estimated by using Ellman and modified by Aykae et al (26,27) similarly inhibition of CCl₄ induced tissue lipid peroxidation was done using the method of Buege and Steven (28).

Effect of pentobarbitone sleeping time

This test was performed to find out the influence on the CCl₄ altered functional performance of the liver. Albino mice of either sex were randomly divided in to 6 groups of 6 animals each. The animals of group I were administered pentobarbitone (40mg/kg) and noted the duration of sleep (duration of loss of righting reflex) was determined. The animals of group II,III, IV, V and VI received ip, CCl₄ (CCl₄:paraffin, 1:1, 4ml/kg), silymarin (100mg/kg), MEMC 200, 400 and 600mg/kg respectively. Half an hour after the respective treatments, animals of group II, III, IV, V and VI were administered pentobarbitone (40mg/kg, ip) and sleeping times were determined. (19)

Statistical analysis

Results were expressed as mean±SE. statistical analysis was performed with one way analysis of variance (ANOVA) followed by Student’s ’t’ test. P value less than 0.05 was considered to be statistically significant when compared to control.

Histopathological observation

Liver tissue collected were used for the preparation of histopathological slides by using microtome and were suitably stained and observed under microscope for the architectural changes seen during CCl₄ challenge in MEMC treated and control groups.
Results

Preliminary phytochemical studies revealed the presence of tannins, alkaloids, phenols, proteins, aminoacids, flavanoids, triterpenoids, β-sitosterol, Vitamins, glycosides. The MEMC found to be non toxic up to 5000 mg/kg.

The antioxidant activity of MEMC in six in vitro models was studied. MEMC demonstrated antioxidant property in all the models. However, MEMC showed better reducing power than the standard i.e. sodium metabisulphite (25 µg). In all other models (i.e DPPH, superoxide, nitric oxide, hydroxyl radical scavenging and inhibition of in vitro lipid peroxidation) antioxidant activity of even 100 µg of MEMC was less than of standards (Table 1).

In the control group (+ve control) CCl₄ significantly enhanced the biochemical markers like SGPT, SGOT, ALP, cholesterol, and bilirubin and reduced the levels of HDL (Table 2). Pretreatment with MEMC (200, 400 and 600 mg/kg) reduced the elevated levels of all the above mentioned biochemical indicators and reduced the liver tissue GSH levels and increased the tissue peroxidation. Pretreatment with MEMC enhanced the tissue GSH levels and reduced lipid peroxidation (Table 3).

Histopathological observations reveal that CCl₄ treatment has damaged the liver architecture (i.e. fatty degeneration of hepatic cells as indicated by ballooning of hepatocytes, fatty cyst, infiltration of lymphocytes and prolifereation of Kupffer cells. Liver sinusoids were congested) and pretreatment with MEMC prevented/reversed the CCl₄ induced liver damage in a dose dependant manner. Fig 1

CCl₄ treatment enhanced the pentobarbitone sleeping time and pretreatment with MEMC reversed it (Fig 2).

The total phenolic content of the MEMC was estimated to be 272.0 ± 2.2 mg Gallic acid equivalents/g of plant extract from triplicate measurements.
### Table 1: *In vitro* antioxidant effect of MEMC.

[Values are mean SE from triplicate readings. Figures in parentheses are % increase (+) and decrease (-) in absorbance]

<table>
<thead>
<tr>
<th>Treatment</th>
<th>Reducing power</th>
<th>Hydroxyl radical scavenging</th>
<th>DPPH scavenging</th>
<th>Nitric oxide scavenging</th>
<th>Superoxide anion scavenging</th>
<th><em>In vitro</em> lipid peroxidation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>0.1270±0.015</td>
<td>0.310±0.012</td>
<td>0.855±0.008</td>
<td>0.292±0.001</td>
<td>0.592±0.003</td>
<td>0.2321±0.014</td>
</tr>
<tr>
<td>CCl₄ 10µl</td>
<td>--</td>
<td>--</td>
<td>--</td>
<td>--</td>
<td>--</td>
<td>0.3617±0.014</td>
</tr>
<tr>
<td>Sodium metabisulphite</td>
<td>0.2382±0.013*</td>
<td>0.071±0.001*</td>
<td>--</td>
<td>--</td>
<td>--</td>
<td>0.0583±0.011*</td>
</tr>
<tr>
<td>(25µg)</td>
<td>(+87.56)</td>
<td>(-77.66)</td>
<td>--</td>
<td>--</td>
<td>--</td>
<td>(-83.88)</td>
</tr>
<tr>
<td>BHT (25µg)</td>
<td>--</td>
<td>--</td>
<td>0.123±0.007*</td>
<td>0.074±0.001*</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td>Vitamin C (25µg)</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>0.0162±0.007</td>
</tr>
<tr>
<td>(25µg)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>(-72.64)</td>
</tr>
<tr>
<td>MEMC (5µg)</td>
<td>0.1723±0.031*</td>
<td>0.241±0.003*</td>
<td>0.701±0.003*</td>
<td>0.233±0.004*</td>
<td>0.469±0.014*</td>
<td>0.2170±0.013*</td>
</tr>
<tr>
<td></td>
<td>(+31.61)</td>
<td>(-22.25)</td>
<td>(-18.01)</td>
<td>(-20.21)</td>
<td>(-20.78)</td>
<td>(-40.01)</td>
</tr>
<tr>
<td>MEMC (10µg)</td>
<td>0.1992±0.021*</td>
<td>0.213±0.007*</td>
<td>0.636±0.006*</td>
<td>0.212±0.007*</td>
<td>0.401±0.003*</td>
<td>0.1975±0.017*</td>
</tr>
<tr>
<td></td>
<td>(+56.85)</td>
<td>(-31.29)</td>
<td>(-25.61)</td>
<td>(-27.4)</td>
<td>(-32.26)</td>
<td>(-45.39)</td>
</tr>
<tr>
<td>MEMC (25µg)</td>
<td>0.2181±0.019*</td>
<td>0.161±0.003*</td>
<td>0.584±0.001*</td>
<td>0.165±0.003*</td>
<td>0.325±0.013*</td>
<td>0.1830±0.020*</td>
</tr>
<tr>
<td></td>
<td>(+71.73)</td>
<td>(-41.40)</td>
<td>(-31.7)</td>
<td>(-43.49)</td>
<td>(-45.1)</td>
<td>(-49.40)</td>
</tr>
<tr>
<td>MEMC (50µg)</td>
<td>0.234±0.032*</td>
<td>0.131±0.006*</td>
<td>0.409±0.009*</td>
<td>0.120±0.006*</td>
<td>0.272±0.012*</td>
<td>0.1623±0.024*</td>
</tr>
<tr>
<td></td>
<td>(+84.33)</td>
<td>(-57.74)</td>
<td>(-52.16)</td>
<td>(-58.9)</td>
<td>(-54.05)</td>
<td>(-55.13)</td>
</tr>
<tr>
<td>MEMC (100µg)</td>
<td>0.2756±0.042*</td>
<td>0.101±0.009*</td>
<td>0.292±0.004*</td>
<td>0.091±0.009*</td>
<td>0.212±0.016*</td>
<td>0.1132±0.025*</td>
</tr>
<tr>
<td></td>
<td>(+117.01)</td>
<td>(-67.42)</td>
<td>(-65.85)</td>
<td>(-68.88)</td>
<td>(-64.19)</td>
<td>(-68.70)</td>
</tr>
</tbody>
</table>

*Significant at P<0.001 compared to control*
Table 2: Effect of MEMC on biochemical parameters of CCl₄-induced hepatic injury.
[Values are mean ± SE from 6 animals in each treatment]

<table>
<thead>
<tr>
<th>Treatment</th>
<th>SGPT (U/l)</th>
<th>SGOT (U/l)</th>
<th>ALP (U/l)</th>
<th>Total bilirubin (mg/dl)</th>
<th>Direct bilirubin (mg/dl)</th>
<th>Cholesterol (mg/dl)</th>
<th>HDL (mg/dl)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control (Saline 0.5ml)</td>
<td>47.51±1.72</td>
<td>0.89±1.11</td>
<td>131.15±1.74</td>
<td>0.92±0.01</td>
<td>0.18±0.01</td>
<td>106.76±2.81</td>
<td>48.41±1.44</td>
</tr>
<tr>
<td>CCl₄</td>
<td>294.31±1.81</td>
<td>479.01±1.72</td>
<td>253.51±1.79</td>
<td>4.25±0.06</td>
<td>1.51±0.04</td>
<td>171.61±4.38</td>
<td>28.59±1.92</td>
</tr>
<tr>
<td>Silymarin (100mg/kg +CCl₄)</td>
<td>61.32±2.30*</td>
<td>134.72±1.76*</td>
<td>93.12±1.97*</td>
<td>1.26±0.04*</td>
<td>0.37±0.02*</td>
<td>120.16±4.83*</td>
<td>46.81±1.24*</td>
</tr>
<tr>
<td>MEMC (200mg/kg +CCl₄)</td>
<td>143.71±0.012*</td>
<td>255.32±1.53*</td>
<td>153.32±1.97*</td>
<td>2.12±0.03*</td>
<td>0.83±0.01*</td>
<td>147.12±4.23*</td>
<td>31.38±1.67*</td>
</tr>
<tr>
<td>MEMC (400mg/kg +CCl₄)</td>
<td>114.32±0.016*</td>
<td>197.91±2.12*</td>
<td>128.1±2.01*</td>
<td>1.86±0.01*</td>
<td>0.67±0.04*</td>
<td>139.31±3.61*</td>
<td>36.71±1.83*</td>
</tr>
<tr>
<td>MEMC (600mg/kg +CCl₄)</td>
<td>87.32±0.06*</td>
<td>163.67±0.91*</td>
<td>93.81±1.191*</td>
<td>1.53±0.01*</td>
<td>0.52±0.01*</td>
<td>126.32±3.82*</td>
<td>41.32±1.70*</td>
</tr>
</tbody>
</table>

*P<0.01 Vs CCl₄ group.
Table 3: Effect of MEMC on the liver tissue levels of GSH and lipid peroxidation in the CCl₄ treated rats.
[Values are mean±SE from 6 animals. Figures are % increase (+) and decrease (-) in absorbance]

<table>
<thead>
<tr>
<th>Group</th>
<th>Tissue levels</th>
<th>Tissue lipid peroxidation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control Saline (0.5ml)</td>
<td>0.9251±0.04</td>
<td>0.2540±0.03</td>
</tr>
<tr>
<td>CCl₄ (2ml/kg)</td>
<td>0.4281±0.06</td>
<td>0.3732±0.05</td>
</tr>
<tr>
<td>Silymarin (100mg/kg +CCl₄)</td>
<td>0.8329±0.04 (+92.46)</td>
<td>0.1332±0.04 (-64.30)</td>
</tr>
<tr>
<td>MEMC (200mg/kg + CCl₄)</td>
<td>0.5994±0.07 (+40.01)</td>
<td>0.2010±0.05 (-46.14)</td>
</tr>
<tr>
<td>MEMC (400mg/kg + CCl₄)</td>
<td>0.6871±0.05 (+60.05)</td>
<td>0.1732±0.04 (-53.59)</td>
</tr>
<tr>
<td>MEMC (600mg/kg +CCl₄)</td>
<td>0.8032±0.04 (+77.80)</td>
<td>0.1320±0.04 (-64.63)</td>
</tr>
</tbody>
</table>

*P<0.01 Vs CCl₄ group
Fig 1: Effect of MEMC on liver architecture in normal and CCl₄ treated rats.

A: Liver architecture of negative control
B: Liver architecture after CCl₄ challenge (+ve control)
C: Liver architecture after Silymarin (100mg/kg)
D: Liver architecture after MEMC (200mg/kg) + CCl₄ challenge
E: Liver architecture after MEMC (400mg/kg) + CCl₄ challenge
F: Liver architecture after MEMC (600mg/kg) + CCl₄ challenge
Fig 2: Effects of MEMC on CCl4 induced prolongation of pentobarbitone sleeping time

A: Pentobarbitone treated (PB), B: CCl4+PB, C: Silymarin+PB, D: MEMC (200mg/kg) + PB, E: MEMC (400mg/kg) + PB, F: MEMC (600mg/kg) + PB.

Discussion

CCl4 induced hepatic damage is due to its cytochrome P-450 enzyme system catalyzed hepatic conversion into highly reactive tricholomethyl radical (CCl3*), which upon reaction with oxygen radical gives trichloromethyl peroxide radical (OOCCl3*). This radical forms covalent bond with sulphydryl group of several membrane molecules like glutathione, which is considered as the initial step in the chain of events leading to lipid peroxidation and hepatic tissue destruction (29-32). Extent of decrease in tissue GSH and tissue lipid peroxidation is a measure of tissue destruction (33) MEMC demonstrated potent DPPH, superoxide, nitric oxide, hydroxyl radical scavenging property. Therefore, it may be inferred that antioxidant property of the study plant may prevent formation of trichloromethylperoxide radical. Thereby inhibit the lipid peroxidation and offer hepatoprotection against CCl4 challenge. Prevention of tissue GSH depletion by MEMC treatment also indicates that the natural inbuilt tissue protective mechanism is kept intact and oxidative degeneration of tissue is prevented. However it is necessary to rule out the inhibitory effect on the cytochrome P-450 enzyme system. The MEMC has exhibited antioxidant property which may be responsible for hepatoprotective activity. The phytochemical test indicated presence of alkaloids, tannins, flavanoids, phenols, glycosides, vitamins, and triterpenes. Some of these constituents have been reported [9, 10], which may responsible for observed activity.
Conclusion

In conclusion, the present study demonstrated that the fruits of Momordica cymbalaria Hook. Fenzl possess hepatoprotective property may be attributed to the antioxidant principles of this plant. Further study is warranted to isolate, characterize and screen active principles from MEMC that possess antioxidant and hepatoprotective properties.

References